

4,6,7,9,10,12,13,15-Octahydro-2*H*-1,3-dithiolo[4,5-*i*][1,4,7,12]dioxadithiacyclo-tetradecine-2-thione

Rui-Bin Hou,^a Bao Li,^b Tie Chen,^a Bing-Zhu Yin^{a*} and Li-Xin Wu^b

^aKey Laboratory of Organism Functional Factors of Changbai Mountain, Yanbian University, Ministry of Education, Yanji 133002, People's Republic of China, and

^bState Key Laboratory of Supramolecular Structure and Materials, College of Chemistry, Jilin University, Changchun 130012, People's Republic of China
Correspondence e-mail: zqcong@jyu.edu.cn

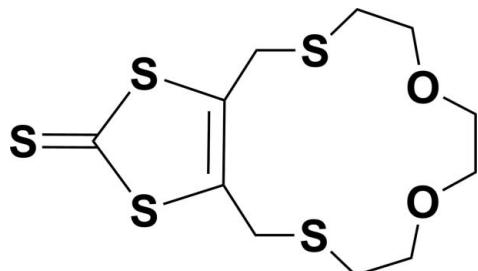
Received 21 July 2009; accepted 24 July 2009

Key indicators: single-crystal X-ray study; $T = 291\text{ K}$; mean $\sigma(\text{C}-\text{C}) = 0.004\text{ \AA}$; R factor = 0.034; wR factor = 0.088; data-to-parameter ratio = 19.6.

In the title molecule, $\text{C}_{11}\text{H}_{16}\text{O}_2\text{S}_5$, the two S atoms from the macrocycle are situated on opposite sides of the mean plane of the five-membered ring, deviating from it by 1.288 (3) and 1.728 (3) \AA . In the crystal, weak intermolecular C–H···S and C–H···O hydrogen bonds link the molecules into layers parallel to (100). The crystal studied was a racemic twin.

Related literature

For crown ether annulated 1,3-dithiol-2-thiones, see: Hansen *et al.* (1992); Trippé *et al.* (2002). For details of the synthesis, see: Chen *et al.* (2005). For a related structure, see: Hou *et al.* (2009)



Experimental

Crystal data

$\text{C}_{11}\text{H}_{16}\text{O}_2\text{S}_5$
 $M_r = 340.54$
 Monoclinic, $P2_1$
 $a = 8.9201 (18)\text{ \AA}$
 $b = 8.5317 (17)\text{ \AA}$
 $c = 10.128 (2)\text{ \AA}$
 $\beta = 97.00 (3)^\circ$
 $V = 765.0 (3)\text{ \AA}^3$

$Z = 2$
 Mo $K\alpha$ radiation
 $\mu = 0.75\text{ mm}^{-1}$

$T = 291\text{ K}$
 $0.13 \times 0.12 \times 0.11\text{ mm}$

Data collection

Rigaku R-AXIS RAPID diffractometer
 Absorption correction: multi-scan (*ABSCOR*; Higashi, 1995)
 $T_{\min} = 0.909$, $T_{\max} = 0.922$

7527 measured reflections
 3221 independent reflections
 3100 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.028$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.034$
 $wR(F^2) = 0.088$
 $S = 1.06$
 3221 reflections
 164 parameters
 1 restraint

H-atom parameters constrained
 $\Delta\rho_{\max} = 0.58\text{ e \AA}^{-3}$
 $\Delta\rho_{\min} = -0.22\text{ e \AA}^{-3}$
 Absolute structure: Flack (1983);
 1359 Friedel pairs
 Flack parameter: 0.42 (9)

Table 1
 Hydrogen-bond geometry (\AA , $^\circ$).

| $D-\text{H}\cdots A$ | $D-\text{H}$ | $\text{H}\cdots A$ | $D\cdots A$ | $D-\text{H}\cdots A$ |
|-----------------------------|--------------|--------------------|-------------|----------------------|
| C7–H7A···S1 ⁱ | 0.97 | 2.86 | 3.695 (3) | 145 |
| C10–H10A···O2 ⁱⁱ | 0.97 | 2.51 | 3.317 (3) | 140 |

Symmetry codes: (i) $x - 1, y, z - 1$; (ii) $-x + 1, y + \frac{1}{2}, -z + 1$.

Data collection: *RAPID-AUTO* (Rigaku, 1998); cell refinement: *RAPID-AUTO*; data reduction: *CrystalStructure* (Rigaku/MSC, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXL97*.

The authors acknowledge financial support from the National Natural Science Foundation of China (grant No. 20662010), the Specialized Research Fund for the Doctoral Program of Higher Education (grant No. 2006184001) and the Open Project of the State Key Laboratory of Supramolecular Structure and Materials, Jilin University.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CV2592).

References

- Chen, T., Liu, W. J., Cong, Z. Q. & Yin, B. Z. (2005). *Chin. J. Org. Chem.* **25**, 570–575.
- Flack, H. D. (1983). *Acta Cryst. A39*, 876–881.
- Hansen, T. K., Jørgensen, T., Stein, P. C. & Becher, J. (1992). *J. Org. Chem.* **57**, 6403–6409.
- Higashi, T. (1995). *ABSCOR*. Rigaku Corporation, Tokyo, Japan.
- Hou, R., Li, B., Yin, B. & Wu, L. (2009). *Acta Cryst. E65*, o1057.
- Rigaku (1998). *RAPID-AUTO*. Rigaku Corporation, Tokyo, Japan.
- Rigaku/MSC (2002). *CrystalStructure*. Rigaku/MSC, The Woodlands, Texas, USA.
- Sheldrick, G. M. (2008). *Acta Cryst. A64*, 112–122.
- Spek, A. L. (2009). *Acta Cryst. D65*, 148–155.
- Trippé, G., Levillain, E., Le Derf, F., Gorgues, A., Sallé, M., Jeppesen, J. O., Nielsen, K. & Becher, J. (2002). *Org. Lett.* **4**, 2461–2464.

supplementary materials

Acta Cryst. (2009). E65, o2042 [doi:10.1107/S1600536809029468]

4,6,7,9,10,12,13,15-Octahydro-2*H*-1,3-dithiolo[4,5-*i*][1,4,7,12]dioxadithiacyclotetradecine-2-thione

R.-B. Hou, B. Li, T. Chen, B.-Z. Yin and L.-X. Wu

Comment

In the context of redox-responsive ligands, TTF is an ideal redox-active unit in view of its unique π -electron donating properties. Attachment of ligands such as crown ethers to TTF in many cases results in the electrochemical tunable ligands (Trippé *et al.*, 2002). Crowned 1,3-dithiole-2-thiones, important precursors to TTF derivatives, have also attracted attention (Hansen *et al.*, 1992). In this paper, we report the crystal structure of the title compound.

In the title compound (Fig. 1), all bond lengths and angles are normal and comparable with those reported for the related structure (Hou *et al.*, 2009). In the crystal, weak intermolecular C—H···S and C—H···O hydrogen bonds (Table 1) link the molecules into layers parallel to $(a+c)b$ plane.

Experimental

The title compound was prepared according to the literature (Chen *et al.*, 2005) and single crystals suitable for X-ray diffraction were prepared by slow evaporation a mixture of dichloromethane and petroleum (60–90 °C) at room temperature.

Refinement

Carbon-bound H-atoms were placed in calculated positions with C—H 0.97 Å and were included in the refinement in the riding model with $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{C})$. The refined value of Flack parameter of 0.42 (9) suggests that the crystal studied was a racemic twin.

Figures

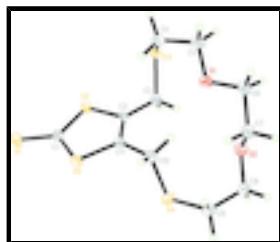


Fig. 1. The molecular structure of the title compound showing the atomic numbering. Displacement ellipsoids of non-H atoms are drawn at the 30% probability level.

4,6,7,9,10,12,13,15-Octahydro-2*H*-1,3-dithiolo[4,5-*i*][1,4,7,12]dioxadithiacyclotetradecine-2-thione

Crystal data

C₁₁H₁₆O₂S₅

$F_{000} = 356$

$M_r = 340.54$

$D_x = 1.478 \text{ Mg m}^{-3}$

Monoclinic, $P2_1$

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

supplementary materials

| | |
|-------------------------------|---|
| Hall symbol: P 2yb | Cell parameters from 7211 reflections |
| $a = 8.9201 (18) \text{ \AA}$ | $\theta = 3.1\text{--}27.5^\circ$ |
| $b = 8.5317 (17) \text{ \AA}$ | $\mu = 0.75 \text{ mm}^{-1}$ |
| $c = 10.128 (2) \text{ \AA}$ | $T = 291 \text{ K}$ |
| $\beta = 97.00 (3)^\circ$ | Block, yellow |
| $V = 765.0 (3) \text{ \AA}^3$ | $0.13 \times 0.12 \times 0.11 \text{ mm}$ |
| $Z = 2$ | |

Data collection

| | |
|---|--|
| Rigaku R-AXIS RAPID diffractometer | 3221 independent reflections |
| Radiation source: fine-focus sealed tube | 3100 reflections with $I > 2\sigma(I)$ |
| Monochromator: graphite | $R_{\text{int}} = 0.028$ |
| $T = 291 \text{ K}$ | $\theta_{\text{max}} = 27.5^\circ$ |
| ω scans | $\theta_{\text{min}} = 3.1^\circ$ |
| Absorption correction: multi-scan (ABSCOR; Higashi, 1995) | $h = -11\text{--}11$ |
| $T_{\text{min}} = 0.909$, $T_{\text{max}} = 0.922$ | $k = -11\text{--}10$ |
| 7527 measured reflections | $l = -13\text{--}11$ |

Refinement

| | |
|--|---|
| Refinement on F^2 | Hydrogen site location: inferred from neighbouring sites |
| Least-squares matrix: full | H-atom parameters constrained |
| $R[F^2 > 2\sigma(F^2)] = 0.034$ | $w = 1/[\sigma^2(F_o^2) + (0.0508P)^2 + 0.1939P]$ where $P = (F_o^2 + 2F_c^2)/3$ |
| $wR(F^2) = 0.088$ | $(\Delta/\sigma)_{\text{max}} = 0.001$ |
| $S = 1.06$ | $\Delta\rho_{\text{max}} = 0.58 \text{ e \AA}^{-3}$ |
| 3221 reflections | $\Delta\rho_{\text{min}} = -0.22 \text{ e \AA}^{-3}$ |
| 164 parameters | Extinction correction: none |
| 1 restraint | Absolute structure: Flack (1983); 1359 Friedel pairs |
| Primary atom site location: structure-invariant direct methods | Flack parameter: 0.42 (9) |
| Secondary atom site location: difference Fourier map | |

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | <i>x</i> | <i>y</i> | <i>z</i> | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|------|-------------|--------------|--------------|----------------------------------|
| C1 | 0.9104 (3) | 0.5403 (4) | 1.0514 (2) | 0.0344 (5) |
| C2 | 0.6930 (2) | 0.4126 (3) | 0.8914 (2) | 0.0264 (5) |
| C3 | 0.5748 (3) | 0.2952 (3) | 0.8447 (2) | 0.0342 (5) |
| H3A | 0.6206 | 0.1920 | 0.8492 | 0.041* |
| H3B | 0.5391 | 0.3164 | 0.7521 | 0.041* |
| C4 | 0.3178 (3) | 0.4733 (4) | 0.8901 (3) | 0.0406 (6) |
| H4A | 0.2524 | 0.4990 | 0.9568 | 0.049* |
| H4B | 0.3929 | 0.5556 | 0.8915 | 0.049* |
| C5 | 0.2242 (3) | 0.4741 (4) | 0.7555 (3) | 0.0445 (7) |
| H5A | 0.1638 | 0.3795 | 0.7439 | 0.053* |
| H5B | 0.1566 | 0.5637 | 0.7482 | 0.053* |
| C6 | 0.2480 (3) | 0.4848 (4) | 0.5262 (3) | 0.0426 (6) |
| H6A | 0.1704 | 0.5650 | 0.5179 | 0.051* |
| H6B | 0.2005 | 0.3842 | 0.5047 | 0.051* |
| C7 | 0.3637 (3) | 0.5190 (4) | 0.4329 (3) | 0.0418 (6) |
| H7A | 0.3150 | 0.5253 | 0.3421 | 0.050* |
| H7B | 0.4128 | 0.6185 | 0.4558 | 0.050* |
| C8 | 0.5855 (3) | 0.4162 (4) | 0.3590 (3) | 0.0427 (7) |
| H8A | 0.5382 | 0.4495 | 0.2722 | 0.051* |
| H8B | 0.6331 | 0.3156 | 0.3480 | 0.051* |
| C9 | 0.7060 (3) | 0.5337 (4) | 0.4090 (3) | 0.0470 (7) |
| H9A | 0.6589 | 0.6351 | 0.4175 | 0.056* |
| H9B | 0.7765 | 0.5438 | 0.3438 | 0.056* |
| C10 | 0.6889 (3) | 0.5696 (4) | 0.6801 (2) | 0.0381 (6) |
| H10A | 0.6885 | 0.6826 | 0.6690 | 0.046* |
| H10B | 0.5861 | 0.5322 | 0.6586 | 0.046* |
| C11 | 0.7448 (2) | 0.5293 (3) | 0.8207 (2) | 0.0290 (5) |
| O1 | 0.3233 (2) | 0.4825 (3) | 0.65686 (18) | 0.0432 (5) |
| O2 | 0.4715 (2) | 0.3969 (2) | 0.44464 (17) | 0.0385 (4) |
| S1 | 1.03598 (9) | 0.58767 (12) | 1.17933 (8) | 0.0558 (2) |
| S2 | 0.78149 (7) | 0.39272 (8) | 1.05411 (6) | 0.03282 (15) |
| S3 | 0.41352 (8) | 0.29178 (9) | 0.93755 (8) | 0.04456 (18) |
| S4 | 0.80970 (8) | 0.48036 (12) | 0.56811 (7) | 0.0539 (2) |
| S5 | 0.89230 (7) | 0.64009 (8) | 0.90156 (7) | 0.03733 (16) |

Atomic displacement parameters (\AA^2)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|----|-------------|-------------|-------------|-------------|--------------|--------------|
| C1 | 0.0240 (11) | 0.0456 (15) | 0.0325 (11) | 0.0034 (10) | -0.0003 (9) | -0.0064 (11) |
| C2 | 0.0195 (10) | 0.0336 (13) | 0.0252 (10) | 0.0033 (8) | -0.0005 (8) | -0.0022 (9) |
| C3 | 0.0310 (12) | 0.0346 (13) | 0.0356 (12) | 0.0012 (10) | -0.0020 (9) | -0.0022 (11) |
| C4 | 0.0330 (13) | 0.0503 (17) | 0.0400 (13) | 0.0006 (12) | 0.0105 (10) | -0.0063 (12) |
| C5 | 0.0259 (12) | 0.0616 (19) | 0.0473 (15) | 0.0072 (12) | 0.0094 (11) | 0.0041 (14) |
| C6 | 0.0275 (13) | 0.0597 (19) | 0.0388 (13) | 0.0049 (12) | -0.0030 (10) | 0.0054 (13) |

supplementary materials

| | | | | | | |
|-----|-------------|-------------|-------------|--------------|--------------|--------------|
| C7 | 0.0369 (14) | 0.0520 (17) | 0.0346 (12) | 0.0011 (12) | -0.0028 (11) | 0.0071 (12) |
| C8 | 0.0417 (15) | 0.0582 (19) | 0.0283 (11) | -0.0032 (13) | 0.0048 (10) | 0.0000 (12) |
| C9 | 0.0413 (15) | 0.069 (2) | 0.0313 (12) | -0.0104 (14) | 0.0065 (11) | 0.0044 (13) |
| C10 | 0.0318 (13) | 0.0507 (16) | 0.0315 (12) | 0.0060 (11) | 0.0028 (10) | 0.0054 (11) |
| C11 | 0.0184 (10) | 0.0390 (13) | 0.0292 (11) | 0.0012 (9) | 0.0020 (8) | -0.0002 (10) |
| O1 | 0.0233 (8) | 0.0716 (14) | 0.0349 (9) | 0.0071 (9) | 0.0045 (7) | 0.0047 (9) |
| O2 | 0.0346 (10) | 0.0433 (11) | 0.0377 (9) | -0.0016 (8) | 0.0049 (7) | 0.0018 (9) |
| S1 | 0.0336 (4) | 0.0887 (7) | 0.0416 (4) | -0.0110 (4) | -0.0092 (3) | -0.0101 (4) |
| S2 | 0.0298 (3) | 0.0407 (3) | 0.0267 (3) | -0.0002 (2) | -0.0015 (2) | 0.0031 (2) |
| S3 | 0.0319 (3) | 0.0508 (4) | 0.0507 (4) | -0.0110 (3) | 0.0041 (3) | 0.0134 (3) |
| S4 | 0.0314 (3) | 0.0975 (7) | 0.0325 (3) | 0.0086 (4) | 0.0027 (3) | -0.0032 (4) |
| S5 | 0.0254 (3) | 0.0466 (4) | 0.0399 (3) | -0.0089 (3) | 0.0031 (2) | 0.0025 (3) |

Geometric parameters (\AA , $^\circ$)

| | | | |
|------------|-------------|------------|-----------|
| C1—S1 | 1.656 (3) | C6—H6A | 0.9700 |
| C1—S2 | 1.708 (3) | C6—H6B | 0.9700 |
| C1—S5 | 1.730 (3) | C7—O2 | 1.413 (4) |
| C2—C11 | 1.341 (4) | C7—H7A | 0.9700 |
| C2—C3 | 1.489 (3) | C7—H7B | 0.9700 |
| C2—S2 | 1.747 (2) | C8—O2 | 1.424 (3) |
| C3—S3 | 1.812 (3) | C8—C9 | 1.511 (4) |
| C3—H3A | 0.9700 | C8—H8A | 0.9700 |
| C3—H3B | 0.9700 | C8—H8B | 0.9700 |
| C4—C5 | 1.510 (4) | C9—S4 | 1.815 (3) |
| C4—S3 | 1.805 (3) | C9—H9A | 0.9700 |
| C4—H4A | 0.9700 | C9—H9B | 0.9700 |
| C4—H4B | 0.9700 | C10—C11 | 1.490 (3) |
| C5—O1 | 1.414 (3) | C10—S4 | 1.824 (3) |
| C5—H5A | 0.9700 | C10—H10A | 0.9700 |
| C5—H5B | 0.9700 | C10—H10B | 0.9700 |
| C6—O1 | 1.409 (3) | C11—S5 | 1.741 (2) |
| C6—C7 | 1.510 (4) | | |
| S1—C1—S2 | 124.03 (17) | C6—C7—H7A | 110.0 |
| S1—C1—S5 | 123.24 (18) | O2—C7—H7B | 110.0 |
| S2—C1—S5 | 112.70 (14) | C6—C7—H7B | 110.0 |
| C11—C2—C3 | 127.4 (2) | H7A—C7—H7B | 108.4 |
| C11—C2—S2 | 115.48 (18) | O2—C8—C9 | 113.8 (2) |
| C3—C2—S2 | 117.08 (18) | O2—C8—H8A | 108.8 |
| C2—C3—S3 | 114.98 (18) | C9—C8—H8A | 108.8 |
| C2—C3—H3A | 108.5 | O2—C8—H8B | 108.8 |
| S3—C3—H3A | 108.5 | C9—C8—H8B | 108.8 |
| C2—C3—H3B | 108.5 | H8A—C8—H8B | 107.7 |
| S3—C3—H3B | 108.5 | C8—C9—S4 | 113.3 (2) |
| H3A—C3—H3B | 107.5 | C8—C9—H9A | 108.9 |
| C5—C4—S3 | 115.9 (2) | S4—C9—H9A | 108.9 |
| C5—C4—H4A | 108.3 | C8—C9—H9B | 108.9 |
| S3—C4—H4A | 108.3 | S4—C9—H9B | 108.9 |
| C5—C4—H4B | 108.3 | H9A—C9—H9B | 107.7 |

| | | | |
|------------|-----------|---------------|-------------|
| S3—C4—H4B | 108.3 | C11—C10—S4 | 110.08 (18) |
| H4A—C4—H4B | 107.4 | C11—C10—H10A | 109.6 |
| O1—C5—C4 | 108.3 (2) | S4—C10—H10A | 109.6 |
| O1—C5—H5A | 110.0 | C11—C10—H10B | 109.6 |
| C4—C5—H5A | 110.0 | S4—C10—H10B | 109.6 |
| O1—C5—H5B | 110.0 | H10A—C10—H10B | 108.2 |
| C4—C5—H5B | 110.0 | C2—C11—C10 | 125.7 (2) |
| H5A—C5—H5B | 108.4 | C2—C11—S5 | 116.22 (18) |
| O1—C6—C7 | 107.8 (2) | C10—C11—S5 | 118.0 (2) |
| O1—C6—H6A | 110.1 | C6—O1—C5 | 113.3 (2) |
| C7—C6—H6A | 110.1 | C7—O2—C8 | 113.1 (2) |
| O1—C6—H6B | 110.1 | C1—S2—C2 | 98.09 (12) |
| C7—C6—H6B | 110.1 | C4—S3—C3 | 103.16 (13) |
| H6A—C6—H6B | 108.5 | C9—S4—C10 | 99.88 (14) |
| O2—C7—C6 | 108.3 (2) | C1—S5—C11 | 97.47 (13) |
| O2—C7—H7A | 110.0 | | |

Hydrogen-bond geometry (Å, °)

| <i>D</i> —H··· <i>A</i> | <i>D</i> —H | H··· <i>A</i> | <i>D</i> ··· <i>A</i> | <i>D</i> —H··· <i>A</i> |
|-----------------------------|-------------|---------------|-----------------------|-------------------------|
| C7—H7A···S1 ⁱ | 0.97 | 2.86 | 3.695 (3) | 145 |
| C10—H10A···O2 ⁱⁱ | 0.97 | 2.51 | 3.317 (3) | 140 |

Symmetry codes: (i) $x-1, y, z-1$; (ii) $-x+1, y+1/2, -z+1$.

supplementary materials

Fig. 1

